



*J. Serb. Chem. Soc.* 75 (8) 1093–1098 (2010)  
JSCS–4034

## On the number of Kekulé structures of fluoranthene congeners

DAMIR VUKIČEVIĆ<sup>1</sup>, JELENA ĐURĐEVIĆ<sup>2</sup> and IVAN GUTMAN<sup>2\*#</sup>

<sup>1</sup>Faculty of Natural Sciences and Mathematics, University of Split, Nikole Tesle 12,  
HR-21000 Split, Croatia and <sup>2</sup>Faculty of Science, University of Kragujevac,  
P. O. Box 60, 34000 Kragujevac, Serbia

(Received 7 December 2009)

**Abstract:** The Kekulé structure count  $K$  of fluoranthene congeners is studied. It is shown that for such polycyclic conjugated  $\pi$ -electron systems, either  $K = 0$  or  $K \geq 3$ . Moreover, for every  $t \geq 3$ , there are infinitely many fluoranthene congeners having exactly  $t$  Kekulé structures. Three classes of Kekuléan fluoranthenes are distinguished: *i*)  $\Phi_0$  – fluoranthene congeners in which neither the male nor the female benzenoid fragment has Kekulé structures, *ii*)  $\Phi_m$  – fluoranthene congeners in which the male benzenoid fragment has Kekulé structures, but the female does not, and *iii*)  $\Phi_{fm}$  – fluoranthene congeners in which both the male and female benzenoid fragments have Kekulé structures. Necessary and sufficient conditions are established for each class,  $\Phi = \Phi_0, \Phi_m, \Phi_{fm}$ , such that for a given positive integer  $t$ , there exist fluoranthene congeners in  $\Phi$  with the property  $K = t$ .

**Keywords:** fluoranthenes; Kekulé structure; polycyclic aromatic hydrocarbons.

### INTRODUCTION

Continuing our studies of the  $\pi$ -electron properties of polycyclic conjugated molecules of the fluoranthene-type,<sup>1–8</sup> in this work, attention was focused on their Kekulé structures. As explained in detail elsewhere,<sup>1,6</sup> the systems considered consist of two benzenoid fragments, joined so as to form an additional five-membered ring (*cf.* Fig. 1). Thus, fluoranthene congeners are, from a structural point of view, closely similar to benzenoid systems.<sup>9–13</sup> The Kekulé structures and various Kekulé-structure-based properties of benzenoid molecules have been extensively studied in the past; for details see the books,<sup>13–16</sup> review<sup>17</sup> and recent papers.<sup>18–21</sup> Analogous properties of the fluoranthene-type systems were, so far, analyzed only to a limited extent.<sup>8</sup> In this paper, we are concerned with the possible number of Kekulé structures (Kekulé structure count,  $K$ ) of fluoranthene congeners. If  $K > 0$ , the respective molecule is said to be Kekuléan.<sup>13,16</sup>

\* Corresponding author. E-mail: gutman@kg.ac.rs

# Serbian Chemical Society member.

doi: 10.2298/JSC091207077V

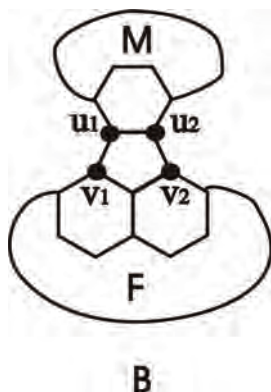


Fig. 1. The general form of a fluoranthene congener **B**; it is obtained by joining two benzenoid fragments (**M** and **F**) so as to form a five-membered ring. The fragments **M** and **F** are referred to as male and female, respectively. For details see text.

Following the terminology of an earlier work,<sup>6</sup> the benzenoid fragments of a fluoranthene-type system are called “male” and “female”, so that the male fragment (**M** in Fig. 1) is connected to the female fragment (**F** in Fig. 1) by two adjacent vertices ( $u_1$  and  $u_2$  in Fig. 1), whereas the female fragment is connected to the male fragment *via* two vertices at distance 2 ( $v_1$  and  $v_2$  in Fig. 1). In other words, three vertices of the five-membered ring belong to the female fragment and two vertices to the male fragment.

Bearing in mind the notation specified in Fig. 1, if **B** is a fluoranthene congener, then by deleting from **B** the edges  $u_1v_1$  and  $u_2v_2$ , a subgraph is obtained consisting of two disconnected benzenoids, referred to as male  $M = M_B$  and female  $F = F_B$ . Denote *i*) by  $\Phi_0$  the class of Kekuléan fluoranthenes in which neither the female nor the male benzenoid fragment have Kekulé structures, *ii*) by  $\Phi_m$  the class of Kekuléan fluoranthenes in which the male fragment has Kekulé structures, but the female does not, and *iii*) by  $\Phi_{fm}$  the class of fluoranthenes in which both the male and female fragments have Kekulé structures.

The main results obtained are the following:

- i*) there exist fluoranthenes of the type  $\Phi_0$  with  $K$  Kekulé structures if and only if  $K$  is the product of two integers both greater than 2;
- ii*) a fluoranthene of type  $\Phi_m$  can have  $K$  Kekulé structures if and only if  $K \geq 3$ ;
- iii*) a fluoranthene of the type  $\Phi_{fm}$  can have  $K$  Kekulé structures if and only if  $K \geq 6$ , and  $K$  is not a prime number, *i.e.*,  $K \neq 7, 11, 13, 17, 19, 23, \dots$

#### NOTATION AND PRELIMINARIES

Let **B** be the molecular graph<sup>22</sup> of a benzenoid or fluoranthene system, and let  $\kappa$  be a Kekulé structure of **B**. Denote by  $d_1 = d_1(\mathbf{B}) = d_1(\mathbf{B}, \kappa)$  the number of double bonds of  $\kappa$  belonging to only one ring and by  $d_2 = d_2(\mathbf{B}) = d_2(\mathbf{B}, \kappa)$  the number of double bonds that are shared by two rings. Let **R** be a ring of **B** and  $D(\mathbf{R}) = D(\mathbf{R}, \kappa)$  the number of double bonds belonging to **R**. Then,

$$\sum_{\mathbf{R}} D(\mathbf{R}) = d_1 + 2d_2$$

where the summation goes over all rings of  $\mathbf{B}$ .

Denote by  $n = n(\mathbf{B})$  the number of vertices of  $\mathbf{B}$ , by  $n_i = n_i(\mathbf{B})$  the number of internal vertices of  $\mathbf{B}$  and by  $h = h(\mathbf{B})$  number of hexagons of  $\mathbf{B}$ . The number of Kekulé structures of  $\mathbf{B}$  is  $K = K(\mathbf{B})$ .

Let  $\mathbf{B}'$  be a subgraph of  $\mathbf{B}$  and let  $\kappa$  be a Kekulé structure of  $\mathbf{B}$ . Then,  $\kappa(\mathbf{B}')$  denotes the set of double bonds of  $\kappa$  that belong to  $\mathbf{B}'$ .

In the proof of the main results, the following theorems were used.

*Theorem A.*<sup>1,16</sup> If  $\mathbf{B}$  is a benzenoid system, then  $n = 4h + 2 - n_i$ . If  $\mathbf{B}$  is a fluoranthene-like system, then  $n = 4h + 5 - n_i$ .

*Theorem B.*<sup>8</sup> *i)* Let  $\mathbf{B} \in \Phi_0$ . Then  $K(\mathbf{B}) = K(F_{\mathbf{B}} - v_i) \cdot K(M_{\mathbf{B}} - u_i)$  for either  $i = 1$  or  $i = 2$ . Moreover, for  $i \neq j$ ,  $K(F_{\mathbf{B}} - v_j) = K(M_{\mathbf{B}} - u_j) = 0$ ; *ii)* let  $\mathbf{B} \in \Phi_m$ . Then  $K(\mathbf{B}) = K(F_{\mathbf{B}} - v_1 - v_2)K(M_{\mathbf{B}})$ . Moreover, if  $\kappa$  is a Kekulé structure of  $\mathbf{B}$ , then  $\kappa(F_{\mathbf{B}} - v_1 - v_2)$  is a Kekulé structure of  $F_{\mathbf{B}} - v_1 - v_2$ ; *iii)* let  $\mathbf{B} \in \Phi_{fm}$ . Then  $K(\mathbf{B}) = K(F_{\mathbf{B}})K(M_{\mathbf{B}})$ .

#### MAIN RESULTS

The main results obtained are stated in the following three theorems.

*Theorem 1.* Let  $\mathbf{B} \in \Phi_0$ . Then  $K(\mathbf{B})$  is equal to the product of two numbers, each greater than 2. Moreover, for every number  $t$  that is a product of two numbers greater than 2, there are infinitely many fluoranthenes  $\mathbf{B} \in \Phi_0$  for which  $K(\mathbf{B}) = t$ .

*Theorem 2.* Let  $\mathbf{B} \in \Phi_m$ . Then  $K(\mathbf{B}) \geq 3$ . Moreover, for every  $t \geq 3$ , there are infinitely many fluoranthenes  $\mathbf{B} \in \Phi_m$  for which  $K(\mathbf{B}) = t$ .

*Theorem 3.* Let  $\mathbf{B} \in \Phi_{fm}$ . Then,  $K(\mathbf{B}) \geq 6$  and  $K(\mathbf{B})$  is not a prime number. For every  $t \geq 6$  which is not a prime number, there is at least one  $\mathbf{B} \in \Phi_{fm}$ , such that  $K(\mathbf{B}) = t$ . Moreover, for every positive integer  $t$  there are infinitely many fluoranthenes with  $t$  Kekulé structures in  $\Phi_{fm}$  if and only if  $t = t_1 t_2 t_3$ , such that  $t_1 \geq 3$ ,  $t_2 \geq 3$  and  $t_3 \geq 2$ .

*Proof of Theorem 1.* Let  $\mathbf{B} \in \Phi_0$  and  $\kappa$  be a Kekulé structure of  $\mathbf{B}$ . It will be shown that  $K(\mathbf{B})$  is the product of two numbers, each of which being greater than 2.

From Theorem B for *i)*, it follows that  $K(\mathbf{B}) = K(M_{\mathbf{B}} - u_i)K(F_{\mathbf{B}} - v_i)$ . Therefore, it is sufficient to prove that  $K(M_{\mathbf{B}} - u_i) \geq 3$  and  $K(F_{\mathbf{B}} - v_i) \geq 3$ . Both proofs are completely analogous and, therefore, only the validity of  $K(F_{\mathbf{B}} - v_i) \geq 3$  will be demonstrated. Note that  $\kappa(F_{\mathbf{B}} - v_i)$  is a Kekulé structure of  $F_{\mathbf{B}} - v_i$ . Furthermore,  $n(F_{\mathbf{B}} - v_i) = n(F_{\mathbf{B}}) - 1$ . Then by Theorem A:

$$d_1(F_{\mathbf{B}}) + d_2(F_{\mathbf{B}}) = \frac{(4h(F_{\mathbf{B}}) + 2 - n_i(F_{\mathbf{B}})) - 1}{2} = 2h(F_{\mathbf{B}}) - \frac{n_i(F_{\mathbf{B}})}{2} + \frac{1}{2}$$

As, evidently,  $d_2(F_{\mathbf{B}}) \geq \frac{1}{2}n_i(F_{\mathbf{B}})$ , one obtains:

$$\sum_{\mathbf{R}} D(\mathbf{R}, \kappa) = d_1(F_{\mathbf{B}}) + 2d_2(F_{\mathbf{B}}) \geq 2h(F_{\mathbf{B}}) + \frac{1}{2}$$

with the left-hand side summation going over all rings of  $F_{\mathbf{B}}$ . Since this sum is an integer, it follows that:

$$\sum_{\mathbf{R}} D(\mathbf{R}, \kappa) \geq 2h(F_{\mathbf{B}}) + 1$$

Hence, at least one hexagon of  $F_{\mathbf{B}}$  contains three double bonds of the Kekulé structure  $\kappa$ .

It is necessary to distinguish between two cases:

*Case 1.* In  $F_{\mathbf{B}}$ , the relation  $D(\mathbf{R}_1, \kappa) = D(\mathbf{R}_2, \kappa) = 3$  holds for (at least) two six-membered rings,  $\mathbf{R}_1$  and  $\mathbf{R}_2$ .

Let  $\kappa_1$  and  $\kappa_2$  be the Kekulé structures obtained by rotation of the double bonds of  $\kappa(F_{\mathbf{B}} - v_i)$  in the rings  $\mathbf{R}_1$  and  $\mathbf{R}_2$ , respectively. Then,  $\kappa_1$ ,  $\kappa_2$  and  $\kappa(F_{\mathbf{B}} - v_i)$  are three distinct Kekulé structures on  $F_{\mathbf{B}} - v_i$ , which means that  $K(F_{\mathbf{B}} - v_i) \geq 3$ .

*Case 2.* In  $F_{\mathbf{B}}$ , the relation  $D(\mathbf{R}, \kappa) = 3$  holds for only one six-membered ring  $\mathbf{R}$ .

Then,  $D(\mathbf{R}') = 2$  for all other six-membered rings  $\mathbf{R}'$ . Let  $\kappa'$  be the Kekulé structure obtained by rotating the double bonds in  $\mathbf{R}$ . Then:

$$\sum_{\mathbf{R}} D(\mathbf{R}, \kappa') \geq 2h(F_{\mathbf{B}}) + 1$$

Hence, either  $\kappa'$  has at least two hexagons  $\mathbf{R}_1$  and  $\mathbf{R}_2$  for which  $D(\mathbf{R}_1, \kappa') = D(\mathbf{R}_2, \kappa') = 3$  or  $D(\mathbf{R}, \kappa') = 3$  and  $D(\mathbf{R}', \kappa') = 2$  for all other hexagons  $\mathbf{R}'$ . In the latter case,  $D(\mathbf{R}', \kappa(F_{\mathbf{B}}))$  and  $D(\mathbf{R}', \kappa')$  would coincide for all hexagons  $\mathbf{R}'$ . However, these two terms differ in the hexagon(s) adjacent to  $\mathbf{R}$ , which is a contradiction. Therefore, it must be  $D(\mathbf{R}_1, \kappa') = D(\mathbf{R}_2, \kappa') = 3$ .

Let  $\kappa_1$  and  $\kappa_2$  be the Kekulé structures obtained by rotation of the double bonds of  $\kappa'$  in the rings  $\mathbf{R}_1$  and  $\mathbf{R}_2$ , respectively. Then,  $\kappa_1$ ,  $\kappa_2$  and  $\kappa'$  are three Kekulé structures of  $F_{\mathbf{B}} - v_i$ , which means that also in this case  $K(F_{\mathbf{B}} - v_i) \geq 3$ .

This proves that  $K(\mathbf{B})$  is a product of two numbers, each of which is greater than 2. The example depicted in Fig. 2 shows that it is possible to design arbitrarily many fluoranthenes of class  $\Phi_0$  for which  $K = t_1 t_2$ ,  $t_1 \geq 3$ ,  $t_2 \geq 3$ .

This concludes the proof of the Theorem.

Proofs of Theorems 2 and 3 are analogous, except that instead of Theorem B (i), it is necessary to use Theorem B (ii) and (iii), respectively.

By combining Theorems 1–3, one obtains:

*Corollary 4.* Let  $\mathbf{B}$  be a Kekuléan fluoranthene. Then  $K(\mathbf{B}) \geq 3$ . Moreover, for every number  $t \geq 3$ , there are infinitely many fluoranthenes  $\mathbf{B}$  such that  $K(\mathbf{B}) = t$ .

According to Theorems 1–3, the minimal Kekulé structure count of a fluoranthene in class  $\Phi_0$ ,  $\Phi_m$  and  $\Phi_{fm}$  is 9, 3 and 6, respectively. The smallest such minimal- $K$  fluoranthenes are depicted in Fig. 3.

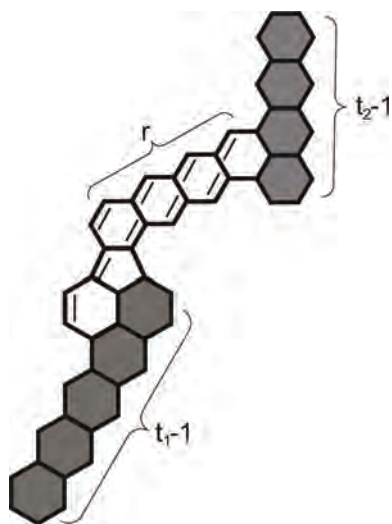


Fig. 2. Fluoranthenes belonging to the class  $\Phi_0$ , having  $t_1 t_2$  Kekulé structures,  $t_1 \geq 3$  and  $t_2 \geq 3$ , irrespective of the value of the parameter  $r$ ,  $r = 1, 2, 3, \dots$ . The fixed double bonds are indicated, whereas the shaded areas are domains in which the  $\pi$ -electrons are delocalized. In this example the shaded areas pertain to polyacenes with  $t_1 - 1$  and  $t_2 - 1$  hexagons; recall that a polyacene with  $h$  hexagons has  $h + 1$  Kekulé structures.

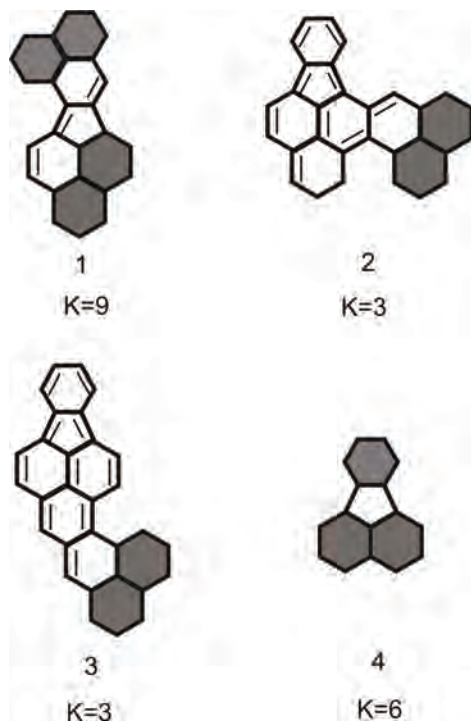


Fig. 3. Fluoranthenes **1**, **2**, **3** and **4** are the smallest members of the classes  $\Phi_0$ ,  $\Phi_m$  and  $\Phi_{fm}$ , respectively, having the smallest possible number of Kekulé structures ( $K$ ). In these formulas the fixed double bonds are indicated, whereas the shaded areas are domains in which the  $\pi$ -electrons are delocalized.

*Acknowledgements.* The partial support of Croatian Ministry of Science, Education and Sport (Grant Nos. 177-0000000-0884 and 037-0000000-2779) is gratefully acknowledged. The authors also thank the Ministry of Science and Technological Development of the Republic of Serbia for partial support of this work through Grant No. 144015G.

## ИЗВОД

## О БРОЈУ КЕКУЛÉ СТРУКТУРА У ЈЕДИЊЕЊИМА ФЛУОРАНТЕНСКОГ ТИПА

DAMIR VUKIČEVIĆ<sup>1</sup>, ЈЕЛЕНА БУРЂЕВИЋ<sup>2</sup> и ИВАН ГУТМАН<sup>2</sup><sup>1</sup>*Faculty of Natural Sciences and Mathematics, University of Split, Nikole Tesle 12, HR-21000 Split, Croatia*  
<sup>2</sup>*Природно-математички факултет Универзитета у Крагујевцу*

Проучаван је број Kekulé структура  $K$  у једињењима флуорантенског типа. Показано је да је код ових полицикличних конјугованих  $\pi$ -електронских система  $K = 0$  или  $K \geq 3$ . Осим тога, за свако  $K \geq 3$  постоји бесконачно много флуорантена са тачно  $K$  Kekulé структура. Разликују се три класе Kekulé флуорантена: *i*)  $\Phi_0$  – флуорантене код који ни мушки ни женски бензеноидни фрагмент немају Kekulé структуре, *ii*)  $\Phi_m$  – флуорантене код којих мушки бензеноидни фрагмент има Kekulé структуре, а женски нема, и *iii*)  $\Phi_{fm}$  – флуорантене код којих и мушки и женски бензеноидни фрагмент имају Kekulé структуре. Одређени су неопходни и довољни услови за сваку класу  $\Phi = \Phi_0, \Phi_m, \Phi_{fm}$ , да за задани позитивни цео број  $m$ , постоји једињење флуорантенског типа у  $\Phi$  са особином  $K = m$ .

(Примљено 7. децембра 2009)

## REFERENCES

1. I. Gutman, J. Đurđević, *MATCH Commun. Math. Comput. Chem.* **60** (2008) 659
2. J. Đurđević, S. Radenković, I. Gutman, *J. Serb. Chem. Soc.* **73** (2008) 989
3. I. Gutman, J. Đurđević, A. T. Balaban, *Polyc. Arom. Comp.* **29** (2009) 3
4. J. Đurđević, I. Gutman, J. Terzić, A. T. Balaban, *Polyc. Arom. Comp.* **29** (2009) 90
5. A. T. Balaban, J. Đurđević, I. Gutman, *Polyc. Arom. Comp.* **29** (2009) 185
6. I. Gutman, J. Đurđević, *J. Serb. Chem. Soc.* **74** (2009) 765
7. I. Gutman, J. Đurđević, S. Radenković, A. Burmudžija, *Indian J. Chem.* **37A** (2009) 194
8. I. Gutman, *Z. Naturforsch.* **65a** (2010) 473
9. N. Trinajstić, *J. Math. Chem.* **5** (1990) 171
10. N. Trinajstić, *J. Math. Chem.* **9** (1992) 373
11. G. Brinkmann, B. Coppens, *MATCH Commun. Math. Comput. Chem.* **62** (2009) 209
12. A. Vesel, *MATCH Commun. Math. Comput. Chem.* **62** (2009) 221
13. I. Gutman, S. J. Cyvin, *Introduction to the Theory of Benzenoid Hydrocarbons*, Springer-Verlag, Berlin, 1989
14. L. Pauling, *The Nature of the Chemical Bond*, Cornell Univ. Press, Ithaca, 1960
15. E. Clar, *The Aromatic Sextet*, Wiley, London, 1972
16. S. J. Cyvin, I. Gutman, *Kekulé Structures in Benzenoid Hydrocarbons*, Springer-Verlag, Berlin, 1988
17. M. Randić, *Chem. Rev.* **103** (2003) 3449
18. I. Gutman, S. Radenković, *J. Serb. Chem. Soc.* **71** (2006) 1039
19. S. Gojak, I. Gutman, S. Radenković, A. Vodopivec, *J. Serb. Chem. Soc.* **72** (2007) 673
20. J. Đurđević, S. Radenković, I. Gutman, *J. Serb. Chem. Soc.* **73** (2008) 989
21. S. Radenković, I. Gutman, *J. Serb. Chem. Soc.* **74** (2009) 155
22. I. Gutman, O. E. Polansky, *Mathematical Concepts in Organic Chemistry*, Springer-Verlag, Berlin, 1986.