J.Serb.Chem.Soc. 66(8) 499–505(2001) JSCS–2879 UDC 547.918 Original scientific paper

The Ferrier rearrangement as the key step in the synthesis of C7–C16-alkyl 2,3-dideoxy glucosides from glucose and C7–C16-alkanols^{1–3}

STANIMIR KONSTANTINOVIĆ^{a#}, JASMINA PREDOJEVIĆ^a, SVETISLAV GOJKOVIĆ^b, VLADIMIR PAVLOVIĆ^{c#} and JÁNOS CSANÁDI^{d#}

 ^aDepartment of Chemistry, Faculty of Science, University of Kragujevac, P. O. Box 60, YU-34000 Kragujevac, E-mail: konstan@eunet.yu, ^bInstitute for Chemistry, Technology and Metallurgy, Njegoševa 12, YU-11000 Belgrade, ^cFaculty of Chemistry, University of Belgrade, P. O. Box 158, YU-11001 Belgrade and ^dInstitute of Chemistry, Faculty of Science, University of Novi Sad, Trg Dositeja Obradovića 3, YU-21000 Novi Sad, Yugoslavia

(Received 8 March 2001)

The Ferrier rearrangement was used as the key step in the synthesis of C₇–C₁₆-alkyl 2,3-dideoxy glucosides from glucose and C₇–C₁₆-alkanols.

Keywords: synthesis of C7-C16-alkyl 2,3-dideoxy glucosides, Ferrier rearrangement.

INTRODUCTION

The Ferrier rearrangement⁴ (Scheme 1) continues to be the pinnacle of the chemistry of the 2,3-unsaturated sugars. The fact that tri-*O*-acetyl glucal⁵ **1** can now be easily synthesized from glycose adds even more to its attractiveness. In the initial version,⁴ the delocalized allyloxocarbenium ion **2** was trapped with an alcohol affording the α -D-product **3**. In the course of his synthetic studies, Frazer-Reed⁶ quenched ion **2** with triethylsilane to give **5** and with a variety of enol silanes to give, for example, **6a**. The concurrent studies of Grynkiewicz and BeMiller⁷ with enol acetates, to give **6b**, for example, were timely developments. The structures of the major anomers of C-glycosides such as **6** follows independently from a comparison of their ¹³C-NMR data, with those of their *cis* (or β -D) counterparts.⁸ However, independent proof comes from the Eschenmoser-Claisen rearrangement⁹ on the axial alcohol **7** which must necesserily give the α -D amide **6c**.¹⁰

RESULTS AND DISCUSSION

In continuation of our investigations¹⁻³ on the synthesis of biodegradable surfactants wich could be obtained from renewable resources, we synthesized C₇–C₁₆-alkyl 2,3-di-

[#] Serbian Chemical Society active member



Scheme 1.

deoxy glucosides from glucose and fatty alcohols using the Ferrier reaction.^{4,11} As depicted in the synthetic Scheme 2, treatment of glucose **8** with acetic anhydride and molten anhydrous sodium acetate at 120 °C gives exclusively the β -D-glucose penta-acetate **9**, which on treatment with hydrogen bromide/acetic acid leads to the glucosyl bromide **10**. Reductive elimination with zink-copper affords tri-*O*-acetyl glucal **1**. Glycosylation with a fatty alcohol ROH (**11**) and boron trifluoride etherate proceeds by a nucleophilic attack at the anomeric center of the glucal **1**. By an allylic rearrangement, the activated ester function at C-3 leaves the molecule to give the alkyl 2,3-unsaturated glucosed $3\alpha/\beta$ (in 85–97 % yield) with the α -anomer largely prevailing ($3\alpha : 3\beta = 10 : 1$). Hydrogenation on paladium on charcoal (Pd/C)proceeded quantitatively affording C7–C16-alkyl 2,3-dideoxy glucosides **12**. For comparison, the reactions of **1** with methanol and 1-butanol were performed.

Following chromatographic purification, the peracetylated products were de-esterified with the mixture methanol : triethylamine : water (2:1:1) to give alkyl α - glucoside 13 α .

For structural assignmenta extensive NMR studies were employed $(^{1}H - ^{1}H)$ homonuclear and $^{1}H - ^{13}C$ heteronuclear chemical correlation spectroscopy (COSY) experiments were performed).

EXPERIMENTAL

General.

Solvents designated as "dry" were distilled prior to use: dichloromethane from calcium hydride; methanol from magnesium. TLC was performed on silica 60 plates GF 254 (Marck). Dry flash



column chromatography was carried out using Merck silica gel 60, particle size 0.04 - 0.06 mm. Spots were visualized by spraying with 10 % sulfuric acid in ethanol (carbohydrates) and 1 % anisaldehyde and 2 % sulfuric acid in glacial acetic acid (noncarbohydrate compounds) and subsequent heating. NMR spectra: Varian Gemini 200 (200 MHz) in CDCl₃ or Bruker 250 (62.9 MHz) in DMSO-d₆. Assignment of the signals was supported by ¹H - ¹H homonuclear and ¹H - ¹³C heteronuclear chemical-shift correlation spectroscopy (COSY) experiments.

The tri-O-acetyl glucal (1) was prepared according to the literature procedure, ^{12,13} via the acetylated sugar 9 and glucosyl bromide 10 (Scheme 2).

Methyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (3a)

General procedure. To a stirred solution of glucal 1 (2.72 g, 9.99 mmol) in dry toluene (25 ml), methanol (11a, 0.40 ml, 10.00 mmol) and a catalytic amount of BF₃ · Et₂O (0.80 ml, 6.50 mmol) were added. The mixture was allowed to react for 1 h and then neutralized by the addition of Na₂CO₃ (2 g). After the solution had been stirred for 30 min, the solids were filtered off and the filtrate was successively washed with a saturated aqueous solution of NaHCO₃ and distilled water. After drying with anhydrous Na₂SO₄, the solvent was evaporated on a rotary evaporator under vacuum affording a syrup (2.28 g). The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3a** in 96 % yield as a 10 : 1 mixture of the α - and β -methyl glucoside. ¹H-and ¹³C-NMR data are given in Tables I and III, respectively.

But-1-yl 4,6-di-O-acetyl-2,3-dideoxy-a-D-erythro-hex-2-enopyranoside (3b)

Tri-O-acetyl glucal (1, 2.72 g, 9.99 mmol) in dry toluene (25 ml) was treated with 1-butanol (11b, 0.90 ml, 9.83 mmol) and a catalytic amount of $BF_3 \cdot Et_2O(0.80 \text{ ml}, 6.50 \text{ mmol})$ as described for **3a**. The described work-up and removal of the solvent gave 2.44 g of a syrupy residue. The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3b** in 85 % yield as a 10 : 1 mixture of the α - and β -but-1-yl glucoside.

Proton	3αα	3b α	3c α, 3d α, 3f α- 3 kα	3εα
H-1	4.94	5.02	5.02-5.03	4.99
Н-2,3	5.85-5.88	5.80-5.97	5.80-5.97	5.77-5.94
H-4	5.29-5.35	5.27-5.34	5.27-5.36	5.27, 5.32
H-5,6	4.03-4.33	4.07-4.31	3.98-4.32	4.0-4.30
H1'	3.46	3.52	3.51	3.33-3.41
H-1b'		3.78	3.78	3.66-3.76
Н-2'		1.52-1.66	1.50-1.70	1.52
OCOOCH ₃	2.09, 2.11	2.09, 2.10	2.08, 2.10	2.04, 2.09
-(CH ₂) _n -		1.33-1.47	1.26-1.30	1.29
-(CH ₂) _n CH ₃		0.94	0.88–0.89	0.90

TABLE I. ¹H-NMR data of the compounds $3a\alpha - 3k\alpha$ (200 MHz, CDCl₃)

TABLE II. ¹H-NMR data of the compounds $12a\alpha - 12k\alpha$ (200 MHz, CDCl₃)

Proton	12a α	12b α	$12c\alpha,12d\alpha,12f\alpha-12k\alpha$	12e α	
H-1	4.73	4.72	4.77-4.82	4.78	
H-2	1.83	1.87	1.92-1.98	1.97	
H-3	1.82	1.73	1.77-1.83	1.82	
H-4	4.73	4.62	4.67-4.73	4.72	
H-5	3.86-3.97	3.80-3.86	3.86-3.97	3.87-3.93	
Н-ба	4.12	4.00	4.05-4.10	4.10	
H-6b	4.27	4.16	4.22-4.27	4.24	
H-1a'	3.37	3.32	3.35-3.41	3.26	
H-1b'		3.60	3.60-3.67	3.58	
Н-2'		1.49	1.55-1.60	1.51	
OCOCH ₃	2.05, 2.09	1.95, 2.00	2.00-2.09	2.05, 2.08	
$-(CH_2)_n-$		1.34	1.21-1.31	1.29	
-(CH ₂) _n CH ₃		0.85	0.83–0.89	0.90	

Hept-1-yl 4,6-di-O-acetyl-2,3-dideoxy-α-D-erythro-hex-2-enopyranoside (3c)

Tri-*O*-acetyl glucal (**1**, 2.72 g, 9.99 mmol) in dry toluene (25 ml) was treated with 1-heptanol (**11c**, 1.40 ml, 9.90 mmol) and a catalytic amount of BF₃ · Et₂O (0.80 ml, 6.50 mmol) as described for **3a**. The desribed work-up and removal of the solvent gave 2.89 g of a syrupy residue. The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3c** in 88 % yield as a 7 : 1 mixture of the α - and β -hept-1-yl glucoside.

Oct-1-yl 4,6-di-O-acetyl-2,3-dideoxy-α-D-erythro-hex-2-enopyranoside (3d)

Tri-O-acetyl glucal (1, 2.72 g, 9.99 mmol) in dry toluene (25 ml) was treated with 1-octanol (11d, 1.6 ml, 10.12 mmol) and a catalytic amount of $BF_3 \cdot Et_2O(0.80 \text{ ml}, 6.50 \text{ mmol})$ as described for **3a**. The described work-up and removal of the solvent gave 3.06 g of a syrupy residue. The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3d** in 87 % yield as a 10:1 mixture of the α - and β -oct-1-yl glucoside.

2-Ethyl-hex-1-yl 4,6-di-O-acetyl-2,3-dideoxy-a-D-erythro-hex-2-enopyranoside (3e)

Tri-*O*-acetyl glucal (1, 2.72 g, 9.99 mmol) in dry toluene (25 ml) was treated with 2-ethyl-1-hexanol (11e, 1.56 ml, 9.98 mmol) and a catalytic amount of BF₃ \cdot Et₂O (0.80 ml, 6.50 mmol) as described for **3a**. The described work-up and removal of the solvent gave 2.94 g of a syrupy residue. The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3e** in 86 % yield as a 13 : 1 mixture of the α - and β -ethylhex-1-yl glucoside.

Non-1-yl 4,6-di-O-acetyl-2,3-dideoxy-a-D-erythro-hex-2-enopyranoside (3f)

Tri-*O*-acetyl glucal (1, 2.72 g, 9.99 mmol) in dry toluene (25 ml) was treated with 1-nonanol (11f, 1.8 ml, 10.32 mmol) and a catalytic amount of BF₃ · Et₂O (0.80 ml, 6.50 mmol) as described for **3a**. The described work-up and removal of the solvent gave 3.79 g of a syrupy residue. The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3f** in 95 % yield as a 10 : 1 mixture of the α - and β -non-1-yl glucoside.

Dec-1-yl 4,6-di-O-acetyl-2,3-dideoxy-α-D-erythro-hex-2-enopyranoside (3g)

Tri-*O*-acetyl glucal (1, 2.72 g, 9.99 mmol) in dry toluene (25 ml) was treated with 1-decanol (11g, 1.9 ml, 9.96 mmol) and a catalytic amount of BF₃ · Et₂O (0.80 ml, 6.50 mmol) as described for **3a**. The described work-up and removal of the solvent gave 3.79 g of a syrupy residue. The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3g** in 89 % yield as a 10 : 1 mixture of the α - and β -dec-1-yl glucoside.

Undec-1-yl 4,6-di-O-acetyl-2,3-dideoxy-a-D-erythro-hex-2-enopyranoside (3h)

Tri-*O*-acetyl glucal (1, 2.72 g, 9.99 mmol) in dry toluene (25 ml) was treated with 1-undecanol (11h, 2.07 ml, 9.97 mmol) and a catalytic amount of $BF_3 \cdot Et_2O$ (0.80 ml, 6.50 mmol) as described for **3a**. The described work-up and removal of the solvent gave 4.05 g of a syrupy residue. The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3h** in 95 % yield as a 7 : 1 mixture of the α - and β -undec-1-yl glucoside.

Dodec-1-yl 4,6-di-O-acetyl-2,3-dideoxy-α-D-erythro-hex-2-enopyranoside (3i)

Tri-O-acetyl glucal (1, 2.72 g, 9.99 mmol) in dry toluene (25 ml) was treated with 1-dodecanol (11i, 1.86 ml, 9.98 mmol) and a catalytic amount of BF₃ · Et₂O (0.80 ml, 6.50 mmol) as described for **3a**. The described work-up and removal of the solvent gave 4.19 g of a syrupy residue. The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3i** in 97 % yield as a 10 : 1 mixture of the α - and β -dodec-1-yl glucoside.

Tetradec-1-yl 4,6-di-O-acetyl-2,3-dideoxy-a-D-erythro-hex-2-enopyranoside (3j)

Tri-*O*-acetyl glucal (1, 1.36 g, 4.99 mmol) in dry toluene (25 ml) was treated with 1-tetradecanol (11j, 1.07 ml, 4.99 mmol) and a catalytic amount of BF₃ · Et₂O (0.40 ml, 3.25 mmol) as described for **3a**. The described work-up and removal of the solvent gave 2.22 g of a syrupy residue. The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3j** in 94 % yield as a 10 : 1 mixture of the α - and β -tetradec-1-yl glucoside.

Hexadec-1-yl 4,6-di-O-acetyl-2,3-dideoxy-α-D-erythro-hex-2-enopyranoside (3k)

Tri-O-acetyl glucal (1, 1.36 g, 4.99 mmol) in dry toluene (13 ml) was treated with 1-tetradecanol (11k, 1.21 ml, 4.99 mmol) and a catalytic amount of $BF_3 \cdot Et_2O$ (0.40 ml, 3.25 mmol) as de-

Compound	C-1	C-2	C-3	C-4	C-5	C-6	C-1'	$(CH_2)_n$	CH ₃
3a α	94.6	128.4	127.2	64.5	66.2	62.2	54.9		
3b α	93.7	128.3	127.5	64.8	66.4	62.5	67.8	18.8, 31.2	13.2
3cα, 3dα, 3fα—3kα	94.1–94.3	128.7–128.9	127–127.9	65.1–65.3	66.6–66.8	62.8-63.0	68.7–69.0	22.3-31.9	13.7–14.0
3εα	94.3	128.5	127.8	65.1	66.7	62.8	71.1	22.7-39.3	10.8, 13.7
12a α	97.4	23.8	28.6	67.7	68.4	63.1	54.5		
12b α	95.8	23.6	28.7	67.6	68.2	63.0	67.4	19.0, 31.2	13.4
12ca, 12da, 12fα-12kα	95.8–96.2	23.7-23.9	28.5–29.0	67.4–67.9	68.2–68.5	62.7-63.2	66.8–67.8	22.2-31.8	13.6–14.0
12e α	96.4	23.9	28.9	67.9	68.5	63.3	67.7	22.9-41.7	10.8, 13.9
13a α	96.5	27.5	29.0	74.6	65.2	61.4	53.8		
13b α	95.0	27.3	29.1	74.5	65.1	61.3	65.5	19.1, 31.3	13.9
13cα, 13dα, 13fα—13kα	95.0–95.1	27.3	29.0–29.1	74.5	65.0–65.1	61.2–61.3	65.8–65.9	22.1-31.3	13.9

TABLE III. 13C-NMR data of the compounds $3a\alpha - 3k\alpha$ (50 MHz, CDCl₃) and $12a\alpha - 12k\alpha$ (50 MHz, CDCl₃), $13a\alpha - 13k\alpha$ (62.5 MHz, DMSO-d₆),

scribed for **3a**. The described work-up and removal of the solvent gave 2.41 g of a syrupy residue. The residue was purified by dry flash chromatography (toluene–ethyl acetate) to afford **3k** in 96 % yield as a 12 : 1 mixture of the α - and β -hexadec-1-yl glucoside.

1-Octyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (12d)

General procedure. A solution of 0.47 g (1.37 mmol) of the unsaturated **3d** in 25 ml of ethyl acetate was hydrogenated in the presence of of 0.25 g of 10 % palladium-on-carbon for 24 h. The catalyst was removed by filtration and the solvent was evaporated to give 0.47 g of a syrup. The residue was purified by dry flash chromatography (toluene-ethyl acetate) to afford **12d** in 97 % yield as a 10 : 1 mixture of the α - and β -octyl glucoside. ¹H and ¹³C NMR data are given in Tables II and III, respectively.

The resulting material was deacetylated by treatment of the purified sample (100–200 mg) with 2 ml of MeOH-Et₃N-H₂O 2 : 1 : 1 at room temperature for 24 h to give the alkyl 2,3-dideo-xy- α -D-erythro-hexopyranoside (13 α).

ИЗВОД

ГЕRRIER-ОВО ПРЕМЕШТАЊЕ КАО КЉУЧНИ СТУПАЊ У СИНТЕЗИ С7–С16-АЛКИЛ-2,3-ДИДЕОКСИ-ГЛУКОЗИДА ИЗ ГЛУКОЗЕ И С7–С16-АЛКАНОЛА

СТАНИМИР КОНСТАНТИНОВИЋ^а, ЈАСМИНА ПРЕДОЈЕВИЋ^а, СВЕТИСЛАВ ГОЈКОВИЋ^б, ВЛАДИМИР ПАВЛОВИЋ^и и ЈАНОШ ЧАНАДИ^д

^аИнсійшіўці за хемціу Природно-майиемайичког факулійейна Универзийнейна у Крагујевцу, й. йр. 60, 34000 Крагујевац, E-mail: konstan@eunet.yu, ^бИнсійшіўці за хемціу, шехнологију и мейиалургију, Његошева 12, 11000 Београд, ^чХемціски факулійейй Универзийнейна у Београду, Сйуденійски йрг 16, й. йр. 158, 11001 Београд и ^оИнсійшійуй за хемціу Природно-майиемайичког факулійнийа Универзийнейна у Новом Саду, Трг Досийнеја Обрадовића 3, 21000 Нови Сад

Ferrier-ово премештање је примењено као кључни ступањ у вишестепеној синтези С7–С₁₆-алкил-2,3-дидеокси-глукозида која полази од глукозе и С7–С₁₆-алканола.

(Примљено 8. марта 2001)

REFERENCES AND NOTES

- 1. S. Konstantinović, J. Predojević, Z. Petrović, A. Spasojević, B. Dimitrijević, G. Milošević, *J. Serb. Chem. Soc.* 64 (1999) 169, and references cited therein
- S. Konstantinović, J. Predojević, V. Pavlović, S. Gojković, J. Csanadi, J. Serb. Chem. Soc. 66 (2001) 65 and references cited therein
- 3. S. Konstantinović, J. Predojević, S. Gojković, V. Pavlović, J. Serb. Chem. Soc. 66 (2001) 73, and references cited therein
- 4. R. J. Ferrier, N. Prasad, J. Chem. Soc. C (1969) 570, 575, and references cited therein
- 5. For the purists: 3,4,6-tri-O-acetyl-1,5-anhydro-2-deoxy-D-arabino-hex-1-enitol
- 6. R. D. Dawe, B. Fraser-Reid, J. Org. Chem. 49 (1984) 522, and references cited therein
- 7. G. Grynkiewicz, J. N. BeMiller, J. Carbohydr. Res. 1 (1982) 121
- 8. O. Achmatowicz, P. Burkowski, Rocz. Chem. 47 (1973) 99
- 9. A. E. Wick, D. Felix, K. Steen, A. Eschenmoser, Helv. Chim. Acta 47 (1964) 2425
- 10. D. B. Tulshian, B. Fraser-Reid, J. Org. Chem. 49 (1984) 518
- 11. R. J. Ferrier, Adv. Carbohydr. Chem. Biochem. 24 (1969) 199, and references cited therein
- 12. B. Iselin, T. Reichstein, Helv. Chim. Acta 27 (1944) 1146 and 1200
- 13. W. Roth, W. Pigman, Methods in Carbohydr. Chem. 2 (1963) 405.

KONSTANTINOVIĆ et al.

504